Conformational Templates for Rational Drug Design: Flexibility of *cyclo*(D-Pro¹-Ala²-Ala³-Ala⁴-Ala⁵) in DMSO Solution

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Long MD simulations (100 ns) for the important model cyclopentapeptide cyclo(D-Pro¹-Ala²-Ala³-Ala⁴-Ala⁵) were performed in explicit DMSO solution using both OPLS-AA and AMBER03 force fields. Simulations revealed conformational transitions between two main conformers, a predominant one (population 93–99%) and a minor conformer (population 0.4–6.7%). These results are in excellent agreement with 20 experimental proton–proton distances estimated for this cyclopentapeptide. The previously discussed γ -turn-like conformation for Ala⁴ was present only in a minor conformer.

Introduction

Because cyclic pentapeptides (CPPs^a) presumably possess limited flexibility in solution, they may serve as convenient conformational templates for studies of ligand—receptor interaction in the rational design of pharmaceuticals. For instance, CPPs may mimic different types of β - and γ -turns, molecular scaffolds of choice in the search for drug candidates inhibiting protein/protein interactions. CPPs can be readily synthesized, are resistant to proteases, and did not provoke immunogenic responses. Conformational features for many CPPs (mostly containing one or two D-amino acid residues) have been extensively studied by NMR measurements and X-ray spectroscopy (see, e.g., ref 2 and references therein).

General approaches proposed to determine 3D structure(s) of CPPs in solution included measuring NMR parameters (e.g., NOEs and vicinal constants) followed by molecular dynamics (MD) simulations employing the experimental NMR parameters as constraints.³ For the important model compound cyclo(D-Pro¹-Ala²-Ala³-Ala⁴-Ala⁵), pA₄, this approach yielded five different 3D structures in DMSO solution,4 two of them containing a conformation close to the γ -turn for the backbone of Ala.⁴ (The conformation of the γ -turn has been first determined as ϕ ca. 70°, ψ ca. -60°. Conformations with the positive ϕ and negative ψ values, such as the γ -turn, are usually considered as forbidden for L-amino acid residues, as well as conformations with negative ϕ and positive ψ values for D-amino acid residues (such as an inverse γ -turn, ϕ ca. -80° , ψ ca. 80°, which is allowed for L-amino acid residues). Indeed, the extensive review of experimental X-ray and NMR studies of 29 model CPPs reported backbone conformations of these types for only two chiral amino acid residues out of 110 indicating the unfavorable energetics of this conformation.² We have, therefore, previously suggested an alternative approach to study conformational flexibility of CPPs.6 The approach estimates statistical weights for low-energy conformations of CPPs determined by independent energy calculations.⁶ For pA₄, we have found that the experimental NMR parameters obtained in Mierke et al.4 were consistent with averaging over five different low-energy structures, none of which contained the γ -turn-like conformation for a L-amino acid residues.⁶ More recent structural studies of novel CPPs focused on antagonists of the CXCR4 receptor also did not report these type of conformations.⁷

Our previous calculations involved, however, some important limitations, such as employment of the ECEPP/2 force field featuring rigid valence geometry with planar nonproline peptide groups (i.e., the corresponding ω angles were fixed at 180°) and the absence of explicit solvent. Also, the very recent NMR study by Heller et al.⁸ re-examined the conformational flexibility of pA₄ in DMSO solution using specific labeling of the backbone CO and NH groups with 13 C and 15 N, respectively.⁸ This study found a minor conformer (15–30%) of pA₄ containing the γ -turn in question.⁸ Our present communication reports the data from the much more thorough computational studies of pA₄ in DMSO solution.

Methods

The molecular dynamics simulations of pA₄ with explicit DMSO solvent molecules were performed using both the OPLS-AA and AMBER03 force field within the GROMACS 3.3 simulation package. A cubic box of volume $2.44 \times 10^4 \text{ Å}^3$ containing 198 DMSO molecules with periodic boundary conditions was used. The OPLS model used for the description of DMSO molecules employed the following set of parameters: r(C-S) = 1.80 Å, r(S-S) = 1.80 ÅO) = 1.53 Å, σ_s = 3.56 Å, σ_o = 2.93 Å, σ_c = 3.81 Å, ϵ_s = 0.395 kcal/mol, $\epsilon_{\rm o} = 0.280$ kcal/mol, $\epsilon_{\rm c} = 0.160$ kcal/mol, $q_{\rm s} = 0.139$ e, $q_{\rm o} = -0.459$ e, and $q_{\rm c} = 0.160$ e (parameters used previously by Zheng and Ornstein¹⁰). This model correctly reproduced the density and ΔH_{vap} for bulk DMSO at 300° K and 1 atm, as follows: density = 1107 kg/m³ and ΔH_{vap} = 52.28 kJ/mol, the experimental values being 1095 kg/m³ and 52.88 kJ/mol, respectively.¹¹ Energies of the solvated peptides were first minimized by 1000 steepest descent steps, and then simulated at 300° K and 1 atm using the constant temperature and pressure algorithm.¹² All MD simulations were performed with a time step of 1 fs and the atomic coordinates were saved every 10 000 steps. The PME algorithm with cutoffs of 13 Å for nonbonded interactions was used during the simulation.

Results and Discussion

Initial MD simulations were performed starting from 10 different conformations of pA_4 found as low-energy structures by preliminary energy calculations employing the ECEPP/2 force field, where all combinations of the local minima of the

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^a Abbreviations: CPP, cyclopentapeptide; NMR, nuclear magnetic resonance; NOE, nuclear Overhauser effect; MD, molecular dynamics; pA₄, cyclo(D-Pro¹-Ala²-Ala³-Ala⁴-Ala⁵); ECEPP, OPLS-AA, and AMBER03, acronyms for force fields.

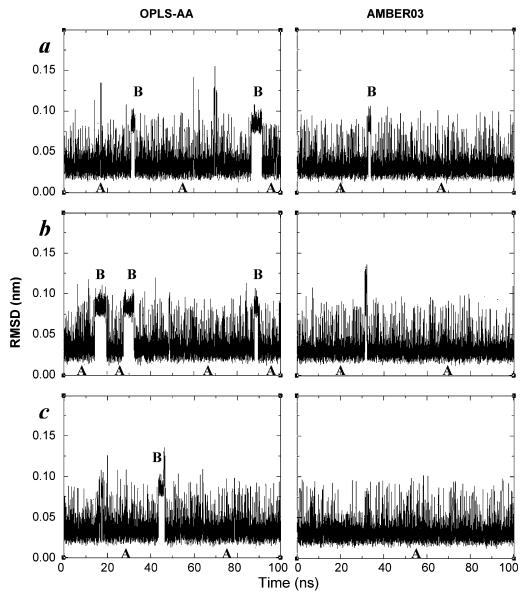


Figure 1. MD trajectories in OPLS-AA force field (left panels) and in AMBER03 force field (right panels) for pA₄ showing root-mean-square deviation values from the initial structure. Panels a, b, and c correspond to runs with different random values for initial velocities. The two different conformational states observed during MD simulation are labeled as A and B.

 ϕ and ψ angles were considered, and the ω angles of the amide bonds were allowed to rotate (see Table S1 in Supporting Information). Initial simulations were run for 20 ns for each of the 10 starting structures. For the OPLS-AA force field, simulations converged to the same (or very similar) single predominant structure in six cases, and in the other four cases, simulations showed transitions between two structures, one of them being the same conformer as the observed predominant structure (data not shown). For the AMBER03 force field, similar results were observed. Additional simulations run for 20 ns for three starting structures obtained by a conformational search using the TINKER package available on the Internet (http://dasher.wustl.edu/tinker/) revealed the same general pattern (see structures 11-13 in Table S1). To analyze conformational equilibrium in pA₄ further, three long MD runs of 100 ns starting from different velocities (randomly generated by GROMACS) were performed for one of the starting conformations where the short run of 20 ns did not show any conformational transitions. The long runs for the OPLS-AA force field (Figure 1, left panels) clearly showed conformational transitions between the same two structures as those found previously in

short runs. Additionally, three additional 100 ns MD simulations were carried out with the AMBER03 force field using the same starting structure. The predominant conformer A and minor conformer B were also sampled during these MD simulations, which agreed with results obtained using the OPLS-AA force field (Figure 1, right panels). It should also be noted that MD trajectories in Figure 1 occasionally featured some narrow peaks corresponding to conformers different from both A and B; because populations of those conformers were very small, they were ignored as insignificant.

According to these MD simulations, the conformational equilibrium of pA₄ in DMSO solution was characterized by transitions between two main conformers determined by the longer MD runs (Figure 1). One of them (conformer A) was a predominant conformer and the other (conformer B) was a minor conformer. Corresponding populations over all trajectories in Figure 1 were about 93% (the OPLS-run trajectories) and about 99.3% (the AMBER03-run trajectories) for conformer A and about 6.7% (OPLS-AA) and about 0.4% (AMBER03) for conformer B. The average values of the dihedral angles for both conformers over trajectories in Figure 1 are listed in Table 1.

Table 1. Dihedral Angles (in Degrees) for Conformers A and Ba

			rmer A value \pm SD	conformer B avg angle value \pm SD		
residue		OPLS-AA	AMBER03	OPLS-AA	AMBER03	
D-Pro ¹	ϕ_I	67.8 ± 9.6	62.7 ± 9.2	69.5 ± 8.3	64.5 ± 9.5	
	ψ_I	-118.9 ± 13.9	-125.3 ± 11.0	-112.7 ± 15.5	-115.2 ± 13.3	
	ω_{12}	176.5 ± 6.2	175.4 ± 6.2	176.4 ± 7.1	177.9 ± 6.8	
Ala ²	ϕ_2	-93.2 ± 17.5	-79.1 ± 12.4	-102.4 ± 23.8	-83.7 ± 19.4	
	ψ_2	8.3 ± 19.0	0.6 ± 15.3	0.2 ± 19.1	-18.5 ± 16.2	
	ω_{23}	177.3 ± 8.8	178.3 ± 9.2	-179.9 ± 8.8	-177.9 ± 9.1	
Ala ³	ϕ_3	-126.5 ± 22.1	-124.5 ± 21.0	-145.5 ± 16.1	-153.8 ± 21.3	
	ψ_3	-119.3 ± 31.8	-127.7 ± 22.1	74.5 ± 28.2	102.0 ± 40.3	
	ω_{34}	173.2 ± 7.4	171.9 ± 7.5	-176.9 ± 6.7	-171.9 ± 9.5	
Ala ⁴	ϕ_4	-85.9 ± 32.9	-74.7 ± 23.8	81.2 ± 9.3	75.5 ± 30.8	
	ψ_4	-32.2 ± 14.6	-30.0 ± 13.2	-48.9 ± 15.5	-47.0 ± 12.3	
	ω ₄₅	159.0 ± 7.8	155.6 ± 8.4	159.3 ± 7.8	160.3 ± 9.6	
Ala ⁵	ϕ_5	-114.7 ± 20.3	-117.9 ± 18.8	-111.5 ± 23.3	-128.2 ± 25.5	
	ψ_5	145.1 ± 11.4	149.6 ± 9.8	137.1 ± 13.7	147.0 ± 12.6	
	ω_{51}	-179.3 ± 8.0	-177.3 ± 8.6	-176.5 ± 7.8	-175.7 ± 8.6	

^a Using the OPLS-AA and AMBER03 force fields averaged over long MD trajectories (Figure 1).

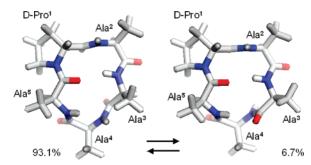


Figure 2. Average structures representing the two conformational states of pA₄, A left) and B (right), which were in equilibrium during 100 ns MD simulation using the OPLS-AA force field. Conformer A does not feature a γ -turn-like conformation for Ala⁴, while conformer B does.

One can see that the predominant conformer A does not include any y-turn-like conformation for the L-amino acid residues, whereas the minor conformer B contains this local conformation for Ala.⁴ Both conformers feature the distinct β -II' turn at D-Pro¹-Ala² stabilized with the hydrogen bond Ala³NH• ··OCAla⁵, with the average N···O distance of 3.18 Å and the average value of NH···O angle of 151.6° (conformer A; the corresponding values for conformer B were 3.20 Å and 165.5°). The γ -turn at Ala⁴ in conformer B was stabilized with the hydrogen bond Ala⁵NH···OCAla³, with the average N···O distance of 3.05 Å and the average value of NH···O angle of 156.3°. At the same time, the average number of the peptide NH groups involved in hydrogen bonds with the SO groups of DMSO along the MD trajectory was 2.14 ± 0.72 , which agrees with the notion that the NH groups of Ala,2 Ala4, and Ala5 interacted with DMSO most of the time. Geometrically, the difference between the two conformers is mostly in orientation of the peptide bond between residues Ala³ and Ala⁴ (see Figure 2). Conformational states A and B observed with the AMBER03 force field possessed very similar structures. In this case, the population of conformer B featuring a γ -turn-like conformation was significantly lower than for that obtained with the OPLS-AA force field.

The longer MD runs in Figure 1 generated average atomatom distances in excellent agreement with the 20 experimental proton-proton distances estimated for pA₄ in DMSO solution by measuring NOEs.⁴ Table 2 lists these data together with the same calculated distances averaged over the trajectories in Figure 1, as well as over fragments of the trajectories corresponding to conformers A and B separately. Averaging over the entire trajectory exactly fit all 20 experimental distances. For pre-

dominant conformer A, no distance was beyond the measured limits for MD runs with both OPLS-AA and AMBER03 force fields. For the minor conformer B (OPLS-AA force field), two distances significantly differed from the experimental limits, namely, $\alpha H_4 - NH_4$ (2.20 \pm 0.11 Å vs limits from 2.46 Å to 2.98 Å) and NH₄-NH₅ (3.32 \pm 0.31 Å vs limits from 2.24 Å to 2.72 Å). It is noteworthy that the differences in protonproton distances αH_3 -NH₄, αH_4 -NH₄, and NH₄-NH₅ are especially indicative of the differences between conformers A and B, while distances $N_3 - C_5$ and $C_3 - N_5$ are almost the same in both conformers (4.04 \pm 0.16 Å and 4.08 \pm 0.17 Å for N₃- C'_{5} and 3.44 \pm 0.28 Å and 3.21 \pm 0.13 Å for C'_{3} -N₅ in conformers A and B, respectively). Slightly different from the conformer obtained with the OPLS-AA force field, minor conformer B obtained with the AMBER03 force field had four distances that significantly differed from the experimental limits, namely, $\alpha H_2 - NH_3$ (3.46 \pm 0.15 Å vs limits from 2.58 Å to 3.12 Å), $\alpha H_3 - NH_4$ (2.16 \pm 0.15 Å vs limits from 2.58 Å to 3.12 Å), $\alpha H_4 - NH_4$ (2.23 \pm 0.07 Å vs limits from 2.46 Å to 2.98 Å), and NH₄-NH₅ (3.47 \pm 0.21 Å vs limits from 2.24 Å to 2.72 Å).

The predominant conformer A, which featured negative values for both ϕ and ψ for Ala,⁴ was somewhat similar to one of the conformers suggested for pA₄ by our previous calculations (see conformer 1 in Table 26). It was not similar, however, to the structure previously proposed as the one with the highest statistical weight in solution.⁶ The negative values of ϕ_4 and ψ_4 were also characteristic for one of the conformers of pA₄ suggested earlier by introducing the experimental NMR parameters as constraints in MD simulations (conformer III⁴). The minor conformer B was not found by our previous calculations; at the same time, similar conformers were represented among structures (conformers I and II) suggested by Mierke et al.4

Several conclusions can be derived from the results of this study. First, the model CPP, pA4, is indeed limited in its conformational flexibility, because unconstrained MD runs starting from very different initial structures all converged to the same two conformers shown in Figure 2. Second, our results clearly showed that averaging over the long unrestricted MD run yielded excellent agreement with available experimental NMR parameters. These results support the general validity of averaging over low-energy conformations independently obtained by energy calculations for CPPs in solution, proposed in our previous study.6 Third, the results on the conformational flexibility of pA₄ were quite similar in MD runs using either

Table 2. Proton-Proton Distances in $Å \pm SD^a$

	exptl limits		total trajectories		conformer A		conformer B	
interproton contact ^b	lower	upper	OPLS-AA	AMBER03	OPLS-AA	AMBER03	OPLS-AA	AMBER03
ProHα-Ala ² NH	2.00	2.37	2.19 ± 0.13	2.18 ± 0.12	2.20 ± 0.13	2.18 ± 0.12	2.21 ± 0.14	2.20 ± 0.12
ProH ^α −Ala ³ NH	3.28	3.97	3.63 ± 0.25	3.57 ± 0.23	3.63 ± 0.26	3.57 ± 0.23	3.67 ± 0.25	3.57 ± 0.23
Proqδ2−Ala ⁵ NH	3.61	4.87	4.79 ± 0.15	4.87 ± 0.13	4.81 ± 0.15	4.87 ± 0.12	4.75 ± 0.14	4.79 ± 0.13
$Proq\delta_2$ -Ala ⁵ H $^{\alpha}$	2.11	3.05	2.65 ± 0.11	2.67 ± 0.12	2.66 ± 0.11	2.67 ± 0.12	2.64 ± 0.11	2.68 ± 0.15
$\text{Proq}\delta_2$ $-\text{Ala}^5\text{q}\beta_3$	2.65	4.60	3.74 ± 0.22	3.63 ± 0.18	3.72 ± 0.21	3.62 ± 0.17	3.84 ± 0.24	3.66 ± 0.21
Ala ² NH−Ala ² H ^α	2.71	2.98	2.95 ± 0.08	2.92 ± 0.08	2.95 ± 0.08	2.92 ± 0.08	2.95 ± 0.08	2.93 ± 0.09
Ala ² NH $-$ Ala ² q β_3	2.56	3.12	2.60 ± 0.17	2.53 ± 0.11	2.59 ± 0.16	2.53 ± 0.11	2.79 ± 0.06	2.56 ± 0.15
Ala ² NH-Ala ³ NH	2.30	2.78	2.58 ± 0.37	2.77 ± 0.29	2.60 ± 0.36	2.77 ± 0.29	2.42 ± 0.33	2.58 ± 0.30
Ala ² Hα-Ala ³ NH	2.58	3.12	3.22 ± 0.24	3.26 ± 0.20	3.20 ± 0.25	3.25 ± 0.20	3.28 ± 0.23	3.46 ± 0.15
$Ala^2q\beta_3-Ala^3NH$	2.76	3.83	3.74 ± 0.22	3.74 ± 0.18	3.75 ± 0.21	3.74 ± 0.18	3.71 ± 0.25	3.51 ± 0.23
Ala ³ Hα−Ala ³ NH	2.44	2.96	2.94 ± 0.09	2.96 ± 0.07	2.94 ± 0.08	2.96 ± 0.07	2.92 ± 0.09	2.91 ± 0.08
Ala ³ Hα-Ala ⁴ NH	2.58	3.12	3.15 ± 0.38	3.22 ± 0.22	3.29 ± 0.21	3.24 ± 0.18	2.41 ± 0.26	2.16 ± 0.15
Ala ³ NH $-$ Ala ³ q β_3	2.66	3.52	2.90 ± 0.21	2.84 ± 0.18	2.88 ± 0.21	2.84 ± 0.18	3.01 ± 0.17	3.07 ± 0.17
Ala ³ NH-Ala ⁵ NH	3.46	4.19	3.58 ± 0.41	3.63 ± 0.38	3.61 ± 0.40	3.64 ± 0.38	3.33 ± 0.36	3.25 ± 0.30
$Ala^4NH-Ala^4H^{\alpha}$	2.46	2.98	2.82 ± 0.28	2.90 ± 0.12	2.92 ± 0.12	2.91 ± 0.09	2.20 ± 0.11	2.23 ± 0.07
Ala ⁴ NH-Ala ⁵ NH	2.24	2.72	2.49 ± 0.46	2.53 ± 0.29	2.34 ± 0.30	2.51 ± 0.27	3.32 ± 0.31	3.47 ± 0.21
Ala^4H^{α} $-Ala^5NH$	3.02	3.64	3.49 ± 0.12	3.49 ± 0.12	3.49 ± 0.12	3.49 ± 0.11	3.47 ± 0.11	3.52 ± 0.10
$Ala^4q\beta_3$ - Ala^5NH	2.61	3.66	3.17 ± 0.23	3.25 ± 0.22	3.18 ± 0.23	3.25 ± 0.22	3.16 ± 0.25	3.24 ± 0.22
Ala ⁵ NH−Ala ⁵ H ^α	2.64	2.98	2.94 ± 0.08	2.95 ± 0.07	2.94 ± 0.08	2.95 ± 0.07	2.94 ± 0.09	2.92 ± 0.08
Ala ⁵ NH $-$ Ala ⁵ q β_3	2.65	3.54	2.80 ± 0.19	2.84 ± 0.16	2.80 ± 0.19	2.84 ± 0.15	2.79 ± 0.20	2.94 ± 0.16

^a Calculated by averaging over trajectories of the 100 ns MD simulations and over two conformational states separately. Experimental measurements taken from Mierke et al.³ Calculated average distances outside experimental range shown in bold. ^b q represents pseudoatoms (i.e., the corresponding $C^β$ or $C^δ$ atoms).

the OPLS-AA or AMBER03 force fields (see Figure 1) showing independence of the force field utilized. Some other specifics of calculation protocols utilized may be more important. For instance, elaborated free-energy calculations applied to pA_4 by others yielded several conformers that violated at least 5 out of the 20 experimental proton—proton distances.¹³

Our results showed that the γ -turn-like conformation for Ala⁴ was present only in the minor conformer B of pA₄, consistent with the rare occurrence of this specific type of conformation for L-amino acid residues in CPPs in available experimental data.^{2,7} In fact, conformer A alone fully satisfied the experimental data of NOE measurements,4 as shown in Table 2. Conformers A and B deduced in this study were very close to the two main conformers suggested very recently by Heller et al. that re-examined the conformational flexibility of pA₄.8 At the same time, this study does not support the assertion that the population of the γ -turn-containing conformer in DMSO could be estimated as high as 15-30%.8 The authors reached that conclusion based primarily on qualitative estimations of ¹³C'-¹H^N cross-peak volumes in long-range HNCO experiments and by MD simulations (40 ns) that employed a protocol identical to that used in this study.⁸ However, they may have used the OPLS-AA parameters for DMSO molecules in GRO-MACS⁹ that do not reflect correctly the bulk properties of DMSO (density = 1067 kg/m³ and ΔH_{vap} = 42.2 kJ/mol, the experimental values being 1095 kg/m³ and 52.88 kJ/mol, respectively, 10 as resulted from our additional calculations using these parameters, which fully reproduced the results of MD simulations by Heller et al.⁸). It should also be noted that our calculations may, in fact, overestimate the population of conformer B in solution, because force fields employing flexible valence geometry generally tend to overestimate the population of the Ramachandran map region with positive ϕ and negative ψ values for L-amino acid residues. ¹⁴ On the other hand, conformers of CPPs similar to conformer B may easily become predominant for CPPs that replace Ala⁴ with glycine, because there are no steric limitations on the γ -turn-like conformations for glycine. 15

Finally, our study established that the conformational flexibility of pA_4 in DMSO solution is almost exclusively limited

to a specific conformer (conformer A). This conformer may be used as a conformational template mimicking, to some extent, different types of β -turns. Specifically, the ϕ , ψ values in Table 1 suggest that the peptide chain reversal at the D-Pro¹-Ala² residues is somewhat close to the β -II′ turn (the standard ϕ , ψ values are 60°, -120° ; -80° , 0°), and the one at Ala⁵-D-Pro¹ may be assigned to the β -V-like turn (the standard ϕ , ψ values are -80° , 80° ; 80° , -80° ; the standard values for the β -turns from Rose et al.¹6). On the other hand, the conformation of Ala⁴ is close to that of the 3/10 helix (the standard ϕ , ψ values were suggested as -57° and $-30^{\circ 17}$).

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Supporting Information Available: Table S1 lists dihedral angles of 13 starting structures of pA₄ and Table S2 lists atomic coordinates (the PDB format) of the averaged structures A and B obtained by employing the OPLS-AA force field. This material is available free of charge via the Internet at http://pubs.acs.org.

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